

Subscriber access provided by ISTANBUL TEKNIK UNIV

# Luffariolides F and G, New Manoalide Derivatives from the Okinawan Marine Sponge Luffariella Sp.

Jun'ichi Kobayashi, Chun-min Zeng, Masami Ishibashi, and Takuma Sasaki

J. Nat. Prod., 1993, 56 (3), 436-439• DOI: 10.1021/np50093a020 • Publication Date (Web): 01 July 2004

Downloaded from http://pubs.acs.org on April 4, 2009

## More About This Article

The permalink http://dx.doi.org/10.1021/np50093a020 provides access to:

- Links to articles and content related to this article
- Copyright permission to reproduce figures and/or text from this article



Chemical Society. 1155 Sixteenth Street N.W., Washington, DC 20036

### LUFFARIOLIDES F AND G, NEW MANOALIDE DERIVATIVES FROM THE OKINAWAN MARINE SPONGE *LUFFARIELLA* SP.

JUN'ICHI KOBAYASHI,\* CHUN-MIN ZENG, MASAMI ISHIBASHI,

Faculty of Pharmaceutical Sciences, Hokkaido University, Sapporo 060, Japan

#### and TAKUMA SASAKI

Cancer Research Institute, Kanazawa University, Kanazawa 920, Japan

ABSTRACT.—Luffariolides F [1] and G [2], two new manoalide-related sesterterpenes with cytotoxic activities, have been isolated from the Okinawan marine sponge *Luffariella* sp. and their structures elucidated on the basis of spectroscopic data.

Manoalide (1)related and sesterterpenes (2-5) have been extensively investigated because of their potent antiinflammatory activity and irreversible inhibition of phospholipase  $A_2$  (6–10). We previously studied a sponge of the genus Luffariella (family Thorectidae; order Dictyoceratida) and isolated five new cytotoxic sesterterpenes, luffariolides A-E(11). Here we describe the isolation and structure elucidation of two other structurally related sesterterpenes, luffariolides F [1] and G [2], both possessing cytotoxic activity, from a different collection of Luffariella sp.

The sponge Luffariella sp. was collected off Kerama Islands, Okinawa and kept frozen until used. The MeOH extract was partitioned between EtOAc and a 1 M NaCl aqueous solution. The EtOAc-soluble fraction was subjected to Si gel flash cc eluted with CHCl<sub>3</sub>/MeOH, followed by purification with reversedphase hplc or gel filtration on a Sephadex LH-20 column, to give luffariolides F[1] (0.0005% yield, wet wt) and G [2] (0.0004%), together with the known sesterterpenes, 6Z-neomanoalide [3a] and 6E-neomanoalide [3b] (1.8:1 mixture, 0.001%) (2), manoalide [4] (0.001%) (1), 6Z-24-acetoxyneomanoalide [5] (0.001%)(4), 6E-neomanoalide-24-al [6] (0.0004%) (4), and (4E,6E)-dehydromanoalide [7] (0.0007%) (5).

Luffariolide F [1] was obtained as a colorless oil, and its ir and uv spectra indicated the presence of an  $\alpha$ ,  $\beta$ -unsatur-



ated ester ( $\nu \max 1740 \text{ cm}^{-1}$ ;  $\lambda \max 210$  nm) and hydroxyl ( $\nu \max 3420 \text{ cm}^{-1}$ ) groups. The molecular formula of **1** was suggested as C<sub>25</sub>H<sub>38</sub>O<sub>5</sub> by the <sup>1</sup>H- and <sup>13</sup>C-nmr spectra together with the hrfabms data (m/z 401.2669 [M-H<sub>2</sub>O+H]<sup>+</sup>,  $\Delta$  -2.3 mmu). The <sup>13</sup>C-nmr spectrum of **1** 



showed signals due to an ester carbonyl, eight olefinic carbons, two sp oxymethines, two sp<sup>3</sup> oxymethylenes, seven sp' methylenes, an sp' quaternary carbon, and four methyls. These data corresponded well to those of 6Zneomanoalide [3a] (2) except for the presence of an additional oxymethine group ( $\delta_c$  70.3, d;  $\delta_H$  3.89, 1H, t, J = 4.4Hz) in 1 and the absence of an sp' methylene group. These differences were clearly due to the presence of a hydroxyl group at C-16. This deduction was supported by the <sup>1</sup>H-<sup>1</sup>H COSY (cross-peaks: H-16/H<sub>a</sub>-17, H-16/H<sub>b</sub>-17, H<sub>a</sub>-17/H<sub>2</sub>-18, and H<sub>b</sub>- $17/H_2$ -18) and the HMBC (12) (crosspeaks: H-16/C-14, H-16/C-15, H-16/C-17, and H-16/C-22) spectra of 1. The H-16 signal appeared as a triplet, being coupled with the methylene protons on C-17 by 4.4 Hz each. This result implied that the H-16 was equatorially oriented, and the hydroxyl group on C-16 was therefore axial. An attempt to determine the absolute stereochemistry of **1** by the modified Mosher method failed due to the failure of **1** to form the Mosher ester and the small amount of material available. The geometries of the  $\Delta^{6.7}$  and  $\Delta^{10,11}$ double bonds were deduced to be 6Z and 10E on the basis of the <sup>13</sup>C chemical shifts of C-24 ( $\delta_{\rm C}$  60.3, t) and C-23 ( $\delta_{\rm C}$  16.1 q) compared with those of 6E- and 6Zneomanoalides (11). The structure of luffariolide F was therefore concluded to be 16-hydroxy-6Z-neomanoalide [**1**].

The ir ( $\nu$  max 3400 and 1740 cm<sup>-1</sup>) and uv ( $\lambda$  max 210 nm) absorptions of luffariolide G [2] indicated that it also possessed butenolide and hydroxyl groups. The <sup>1</sup>H- and <sup>13</sup>C-nmr data in combination with the hrfabms results  $(m/z 401.2693 [M-H_2O+H]^+, \Delta + 0.1$ mmu) suggested the molecular formula of 2 to be  $C_{25}H_{38}O_5$ , being the same as that of luffariolide F [1]. The eims of 2 showed an intense peak at m/z 137, implying the presence of the alkylated cyclohexenyl end group commonly generated by manoalide-related sesterterpenes (8). The  $^{1}$ H- and  $^{13}$ C-nmr data of 2 were mostly parallel to those of 6Zneomanoalide [3a] (2). The structural differences between 2 and 3a were found in the C-9-C-11 part of the molecule. The DEPT experiment of 2 revealed the presence of an oxygenated sp<sup>2</sup> quaternary carbon ( $\delta_c$  73.7, s) which was assigned to C-11, bearing a tertiary methyl and a tertiary hydroxyl group as evidenced by the HMBC correlation between C-11 and methyl protons on C-23. The HMBC spectrum of **2** also showed a cross-peak from an olefinic proton at  $\delta_{\rm H}$  5.59 (H-10) to C-11. In the <sup>1</sup>H-<sup>1</sup>H COSY spectrum of **2**, coupling between H-10 and H-9 ( $\delta_{\rm H}$ 5.65;  $J_{9,10}$  = 16.0 Hz, E configuration) was evident. H-9 in turn coupled to the methylene protons on C-8 ( $\delta_{\rm H}$  2.84 and 2.76, each 1H, dt, J=15.0 and 6.5 Hz). From these observations an E double bond was shown to be present at the C-9

Cell Line	Compound						
	1	2	3*	4	5	6	7
L1210 KB	1.6 33.4% <sup>⊾</sup>	2.8 30.8% <sup>⊾</sup>	1.7 36.9%⁵	0.032 0.31	3.0 44.8% <sup>⊾</sup>	1.0 5.6	0.45 2.1

TABLE 1. Cytotoxic Activities of Compounds 1-7 (IC<sub>50</sub> values, µg/ml).

<sup>1</sup>1.8:1 mixture of **3a** and **3b**.

<sup>b</sup>Inhibition (%) at 10 µg/ml.

and C-10 position. Thus the structure of luffariolide G was assigned as **2**.

Cytotoxic activities of luffariolides F [1] and G [2] together with those of known compounds 3–7 against murine lymphoma L1210 cells and human epidermoid carcinoma KB cells in vitro were examined and are shown in Table 1.

#### EXPERIMENTAL

GENERAL METHODS.—The ir and uv spectra were recorded on JASCO A-102 and Shimadzu UV-220 spectrophotometers, respectively. <sup>1</sup>H- and <sup>13</sup>C-nmr spectra were recorded on JEOL GX-270 and EX-400 spectrometers. Fab mass spectra were obtained on a JEOL HX-110 spectrometer using 2-nitrobenzylalcohol as matrix. Ei mass spectra were recorded on a JEOL DX-303 spectrometer. Wako C-300 Si gel was used on glass cc, and tlc was carried out on Merck Si gel GF<sub>254</sub>.

SPONGE MATERIAL.—The sponge Luffariella sp. (order Dictyoceratida; family Thorectidae), collected by scuba off Kerama Islands, Okinawa, was kept frozen until used. The specimen has dark yellow-brown conulose surface and light yellowbrown interior. The mesophyll is dense; the sponge is firm and slightly compressible. Primary and secondary skeletal fibers are the same size; the tertiary skeletal fibers are finer. The primary fibers are 55 µm wide and uncored. The fibers are stratified. The voucher specimen (SS-245) was deposited at the Faculty of Pharmaceutical Sciences, Hokkaido University. This Luffariella sponge (SS-245) is apparently a different species from the Luffariella sponge (SS-15) of our previous study (11). The contents of the sesterterpenes in these two Luffariella sponges are also different; e.g., the latter did not contain manoalide [4].

ISOLATION.—The MeOH extract of the sponge (1.3 kg wet wt) was evaporated under reduced pressure, and the residue (40 g) was partitioned between EtOAc (400 ml $\times$ 3) and 1 M

NaCl (400 ml). The EtOAc-soluble material (4.6 g) was partially (3.8 g) subjected to Si gel flash cc with gradient elution of MeOH in CHCl,  $(0\rightarrow 100\%)$ . The fraction (216 mg) eluted with 10% MeOH in CHCl, was separated by the second Si gel column eluted with  $0 \rightarrow 50\%$  MeOH in CHCl<sub>3</sub>. The fraction (14 mg) eluted with 8% MeOH in CHCl, was further purified by a Sephadex LH-20 column (50% MeOH in CHCl,) to give luffariolide F [1] (7.0 mg). The fraction (36.1 mg) of the second Si gel column eluted with 10% MeOH in CHCl, was subjected to a Sephadex LH-20 column (50% MeOH in CHCl.), followed by reversed-phase hplc (YMC-Pack AM-323; 10×250 mm; 75% MeOH in H2O; flow rate 2.0 ml/min) to give luffariolide G [2] (5.4 mg).

Luffariolide F [1].—A colorless oil:  $[\alpha]^{19}$ D  $-5.9^{\circ}$  (c=0.67, MeOH); uv (MeOH)  $\lambda$  max 210 nm (€ 15000); ir (KBr) v max 3420, 2950, 1740, 1440 cm<sup>-1</sup>; <sup>1</sup>H nmr (CDCl<sub>3</sub>) δ 6.02 (1H, br s, H-2), 5.24 (1H, dd, J=7.6 and 7.0 Hz, H-6) 5.08-5.11 (2H, m, H-10 and H-4), 4.50-4.53 (2H, m,  $H_2$ -25), 4.12(2H, d, J=2.9 Hz,  $H_2$ -24), 3.89(1H, t, J=4.4 Hz, H-16), 2.79 (1H, m, H-5), 2.55 (1H, m, H, -5), 2.12-2.17 (4H, m, H, -8 and H, -9), 2.03 (4H, brs, H,-12 and H,-13), 1.74 (3H, s, H,-22), 1.63 (3H, s, H<sub>3</sub>-23), 1.80 (1H, m, H<sub>4</sub>-17), 1.65 (1H, m, H<sub>b</sub>-17), 1.43 (2H, m, H<sub>2</sub>-18), 1.04 (3H, s, H<sub>3</sub>-20), 0.97 (3H, s, H<sub>3</sub>-21); <sup>13</sup>C nmr (CDCl<sub>3</sub>) δ 172.4 (s, C-1), 116.1 (d, C-2), 171.1 (s, C-3), 81.7 (d, C-4), 30.3 (t, C-5), 120.0 (d, C-6), 143.3 (s, C-7), 35.6 (t, C-8), 28.0 (t, C-9), 123.3 (d, C-10), 136.4, (s, C-11), 39.7 (t, C-12), 26.7 (t, C-13), 142.2 (s, C-14), 128.7 (s, C-15), 70.3 (d, C-16), 32.7 (t, C-17), 34.6 (t, C-18), 35.4 (s, C-19), 28.6 (s, C-20), 27.0 (s, C-21), 16.9 (q, C-22), 16.1 (q, C-23), 60.3 (t, C-24), 58.7 (t, C-25) ppm; eims m/z (rel. int.)  $[M-H_2O]^+$  400 (4), 203 (9), 147 (18), 135 (100); hrfabrns m/z 401.2669 (calcd for  $C_{1}H_{1}O_{1}[M-OH]^{+} 401.2692).$ 

Luffariolide G [2].—A colorless oil:  $[\alpha]p^{24}$ -9.5° (c=0.2, MeOH); uv (MeOH)  $\lambda$  max 210 nm ( $\epsilon$  8300); ir (KBr)  $\nu$  max 3400, 2950, 1740 cm<sup>-1</sup>; <sup>1</sup>H nmr (C<sub>6</sub>D<sub>6</sub>)  $\delta$  5.79(1H, d, J=1.5 Hz, H- 2), 5.65 (1H, dt, J=16.0 and 6.5 Hz, H-9), 5.59 (1H, d, J=16.0 Hz, H-10), 5.07 (1H, m, H-6),4.46(1H, m, H-4), 3.97(2H, s, H, -24), 3.82(1H,  $d, J = 16.6 Hz, H_{2} - 25), 3.72 (1H, d, J = 16.6 Hz,$  $H_{b}$ -25), 2.84 (1H, dd, J=15.0 and 6.5 Hz,  $H_{a}$ -8), 2.76 (1H, dd, J=15.0 and 6.5 Hz, H<sub>b</sub>-8), 2.42  $(1H, m, H_{2}-5), 2.26(2H, m, H_{2}-13), 2.04(1H, m, H_{2}-13))$  $H_{b}$ -5), 1.96 (2H, m,  $H_{2}$ -13), 1.72 (3H, s,  $H_{3}$ -22), 1.71 (2H, m, H<sub>2</sub>-12), 1.63 (2H, m, H<sub>2</sub>-17), 1.51 (2H, m, H<sub>2</sub>-18), 1.31 (3H, s, H<sub>3</sub>-23), 1.17 (6H, s, H<sub>3</sub>-20 and H<sub>3</sub>-21); <sup>13</sup>C nmr (CDCl<sub>3</sub>) δ 172.5 (s, C-1), 116.2 (d, C-2), 171.6 (s, C-3), 81.5 (d, C-4), 30.2 (t, C-5), 120.5 (d, C-6), 142.3 (s, C-7), 38.2 (t, C-8), 125.5 (d, C-9), 139.0 (d, C-10), 73.7 (s, C-11), 42.7 (t, C-12), 22.9 (t, C-13), 136.6 (s, C-14), 127.1 (s, C-15), 32.7 (t, C-16), 19.5 (t, C-17), 38.9 (t, C-18), 34.9 (s, C-19), 28.6 (q, C-20), 28.6 (q, C-21), 19.8 (q, C-22), 27.4 (q, C-23), 60.1 (t, C-24), 58.5 (t, C-25) ppm; eims m/z (rel. int.)  $[M-H_{2}O]^{+}$  400 (4), 385 (3), 367 (3), 287 (6), 137 (100); hrfabms m/z 401.2693 (calcd for C<sub>25</sub>H<sub>37</sub>O<sub>4</sub>  $[M-OH]^+$  401.2692).

#### ACKNOWLEDGMENTS

We thank Mr. Z. Nagahama, Onna, Okinawa, for his help with sponge collection and Dr. J. Fromont, James Cook University of North Queensland, Townsville, Australia, for identification of the sponge. This work was partly supported by a Grant-in-Aid for Scientific Research from the Ministry of Education, Science, and Culture of Japan.

#### LITERATURE CITED

- E.D. de Silva and P.J. Scheuer, *Tetrahedron* Lett., 21, 1611 (1980).
- E.D. de Silva and P.J. Scheuer, *Tetrahedron* Lett., 22, 3147 (1981).
- M.R. Kernan D.J. Faulkner, and R.S. Jacobs, J. Org. Chem., 52, 3081 (1987).
- G.M. König, A.D. Wright, and O. Sticher, J. Nat. Prod., 55, 174 (1992).
- B.C.M. Potts, R.J. Capon, and D.J. Faulkner, J. Org. Chem., 57, 2965 (1992).
- F. Ghomashchi, B.-Z. Yu, E.D. Mihelich, M.K. Jain, and M.H. Gelb, *Biochemistry*, 30, 9559 (1991).
- P.B. Jacobson, L.A. Marshall, A. Sung, and R.S. Jacobs, *Biochem. Pharmacol.*, **39**, 1557 (1990).
- L.J. Reynolds, B.P. Morgan, G.A. Hite, E.D. Mihelich, and E.A. Dennis, *J. Am. Chem. Soc.*, **110**, 5172 (1988).
- C.F. Bennett, S. Mong, M.A. Clarke, L.I. Kruse, and S.T. Crooke, *Biochem. Pharmacol.*, 36, 733 (1987).
- 10. K.B. Glaser and R.S. Jacobs, *Biochem. Pharmacol.*, **36**, 2079 (1987).
- M. Tsuda, H. Shigemori, M. Ishibashi, T. Sasaki, and J. Kobayashi, *J. Org. Chem.*, **57**, 3503 (1992).
- A. Bax and M.F. Summers, J. Am. Chem. Soc., 108, 2093 (1986).

Received 3 September 1992